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**UTILITY
PATENT APPLICATION
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Only for new nonprovisional applications under 37 CFR § 1.53(b))

Attorney Docket No.	98U020	
First Inventor or Application Number	Whiteker, et al.	
Title	Olefin Polymerization Catalysts, Their Production and Use	
Express Mail Label No.	EM354493284US	

APPLICATION ELEMENTS

1. Fee Transmittal Form (e.g., PTO/SB/17)
(Submit an original, and a duplicate for fee processing)

2. Specification Total Pages 52
(preferred arrangement set forth below)
 - Descriptive title of the Invention
 - Cross References to Related Applications (if any)
 - Background of the Invention
 - Brief Summary of the Invention
 - Brief Description of the Drawings (if filed)
 - Detailed Description
 - Claim(s)
 - Abstract of the Disclosure

3. Drawing(s) (35 U.S.C. 113) Total Sheets

4. Oath or Declaration Total Pages

a. Newly executed (original or copy)

b. Copy from a prior application (37 C.F.R. § 1.63(d))
(for continuation/division with Box 15 completed)
[Note Box 5 below]

i. DELETION OF INVENTOR(S)
Signed statement attached deleting
inventor(s) named in the prior application,
see 37 C.F.R. §§ 1.63(d)(2) and 1.33(b).

15. If a CONTINUING APPLICATION, check appropriate box, and supply the requisite information below and in a preliminary amendment

Continuation Divisional Continuation-in-part (CIP)

of prior applications No: _____, from which priority is claimed under 35 U.S.C. § 120

Prior application information: Examiner _____ Group Art Unit _____

16. If a NONPROVISIONAL APPLICATION based upon a foreign or U.S. provisional application, priority is claimed from applications

No. _____, country _____, filed on _____.

17. CORRESPONDENCE ADDRESS

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APPLICATION OF U.S. PATENT

Title: Olefin Polymerization Catalysts, Their Production and Use.

5 Inventors: Gregory T. Whiteker and Jack A. Smith

Field of the Invention

10 This invention relates to a new family of olefin polymerization catalysts based upon phenoxide complexes of transition metals.

Background of the Invention

15 The intense commercialization of metallocene polyolefin catalysts (metallocene being cyclopentadienyl based transition metal catalyst compounds) has led to widespread interest in the design of non-metallocene, homogeneous catalysts. This field is more than an academic curiosity as new, non-metallocene catalysts may provide an easier pathway to currently available products and may also provide product and process opportunities which are beyond the capability of metallocene catalysts. In addition, 20 certain non-cyclopentadienyl ligands will be more economical due to the relative ease of synthesis of a variety of substituted analogs.

25 Anionic, multidentate heteroatom ligands have received the most attention in non-metallocene polyolefins catalysis. Notable classes of bidentate anionic ligands which form active polymerization catalysts include $\text{N}-\text{N}^-$ and $\text{N}-\text{O}^-$ ligand sets. Examples of these types of non-metallocene catalysts include amidopyridines (Kempe, R., "Aminopyridinato Ligands - New Directions and Limitations", 80th Canadian Society for Chemistry Meeting, Windsor, Ontario, Canada, June 1-4, 1997. Kempe, R. et al, *Inorg. Chem.* 1996 vol 35 6742.) Likewise, recent reports by Jordan

et al. of polyolefin catalysts based on hydroxyquinolines (Bei, X.; Swenson, D. C.; Jordan, R. F., *Organometallics* 1997, 16, 3282) have been interesting even though the catalytic activities of Jordan's hydroxyquinoline catalysts is low.

5 European Patent Application 0 803 520 discloses polymerization catalysts containing beta-diketiminate ligands. Other recent non-metallocene olefin polymerization catalysts include US 4,057,565 which discloses 2-dialkylaminobenzyl and 2-dialkylaminomethylphenyl derivatives of selected transition metals and WO 96/08498 which discloses group 4 metal complexes containing a bridged non-aromatic, anionic dienyl ligand group.

10 US 5,637,660 discloses bidentate pyridine based transition metal catalysts.

15 Further Grubbs et al in *Organometallics*, Vol 17, 1988 page 3149-3151 disclose that nickel (II) salicylaldiminato complexes combined with $B(C_6F_5)_3$ polymerized ethylene. (49,500 Mw, Mw/Mn 6.8, and 35 branches per 1000 C's).

20 Ethylenebis(salicylideneiminato)zirconium dichloride combined with methyl alumoxane deposited on a support and unsupported versions were used to polymerize ethylene by Repo et al in *Macromolecules* 1997, 30, 171-175.

Further EP 241,560 A1 discloses alkoxide ligands in transition metal catalyst systems.

25 Thus there is a need in the art for new novel olefin polymerization catalysts.

Summary of the Invention

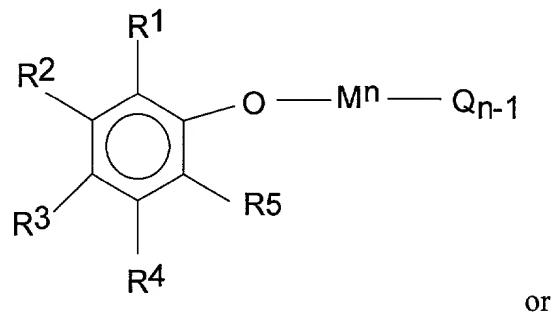
This invention relates to a catalyst system comprising an activator and one or more heteroatom substituted phenoxide group 4 to 10 or lanthanide transition metal compounds wherein the metal is bound to the oxygen of the phenoxide group and provided that:

- 5 a) if more than one heteroatom substituted phenoxide is present it is not bridged to the other heteroatom substituted phenoxide,
- b) if the metal is a group 4 metal, the heteroatom substituted phenoxide does not contain pyridine,
- c) if the metal is a group 4 metal then the carbon ortho to the carbon bound to the oxygen of the phenoxide may not be bound to an aldehyde or an ester, and
- 10 d) if the metal is nickel then the carbon ortho to the carbon bound to the oxygen of the phenoxide may not be bound to an imine.

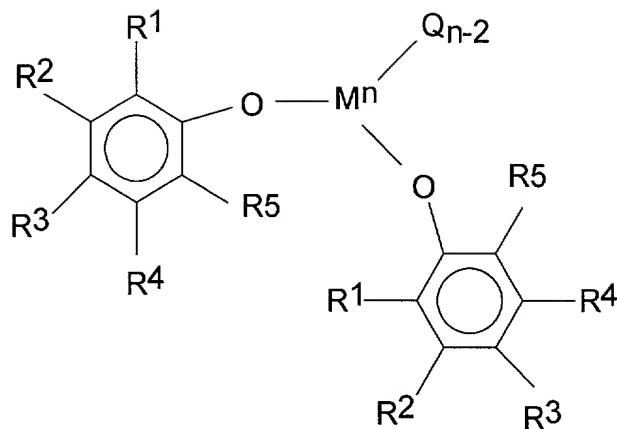
The activator is preferably one or more of aluminum alkyl, an alumoxane, a modified alumoxane, a non-coordinating anion, or a borane.

15

This invention further relates to a novel olefin polymerization systems comprising an activator and one or more catalysts represented by the of the following formulae:



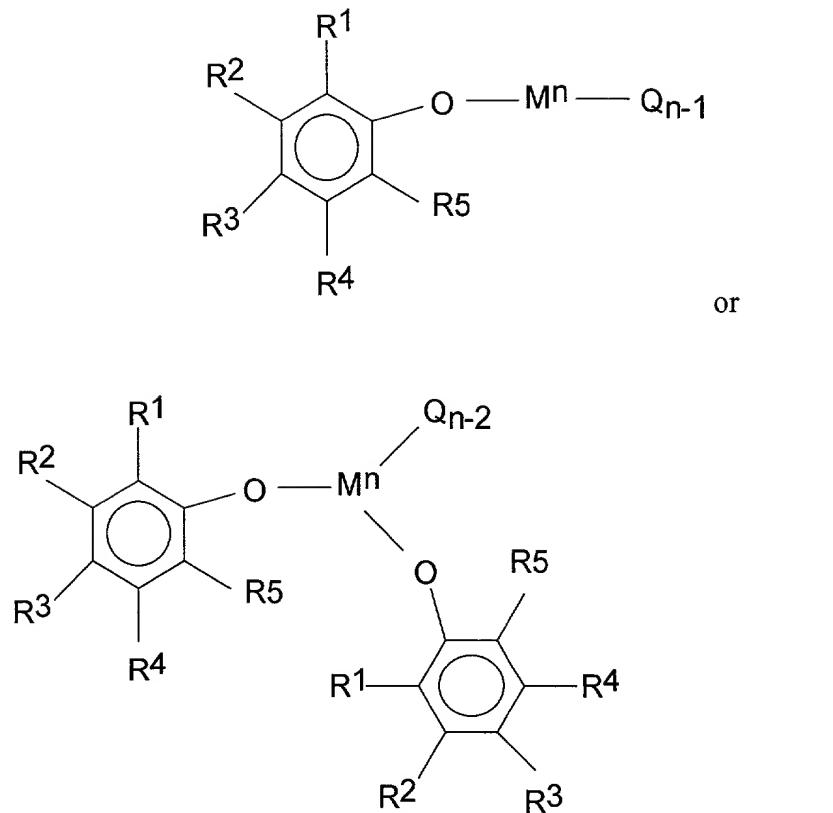
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wherein R¹ to R⁵ may be independently hydrogen, a heteroatom containing group or a C₁ to C₁₀₀ group provided that one of R² to R⁵ is a group containing a heteroatom (R⁵ and/or R¹ also may or may not be bound to the metal M), and further provided that the R⁴ and R⁵ groups do not form pyridine in the first formula if M is a group 4 metal and the R⁴ and R⁵ groups do not form pyridine in at least one ring of the second formula if M is a group 4 metal, O is oxygen, M is a group 4 to 10 transition metal or lanthanide metal, n is the valence state of M, Q is an anionic ligand or a bond to an R group containing a heteroatom which may be any of R¹ to R⁵, and further provided that if M is a group 4 metal then R⁵ may not be an aldehyde or an ester, and further provided that if M is nickel then R⁵ may not be an imine. Any two or more R groups may form a ring structure. The activator is preferably an aluminum alkyl, an alumoxane, a modified alumoxane, a non-coordinating anion, a borane or a combination thereof.

Detailed Description of the Invention

This invention relates to a novel olefin polymerization system comprising an activator and one or more catalysts represented by the following formulae:

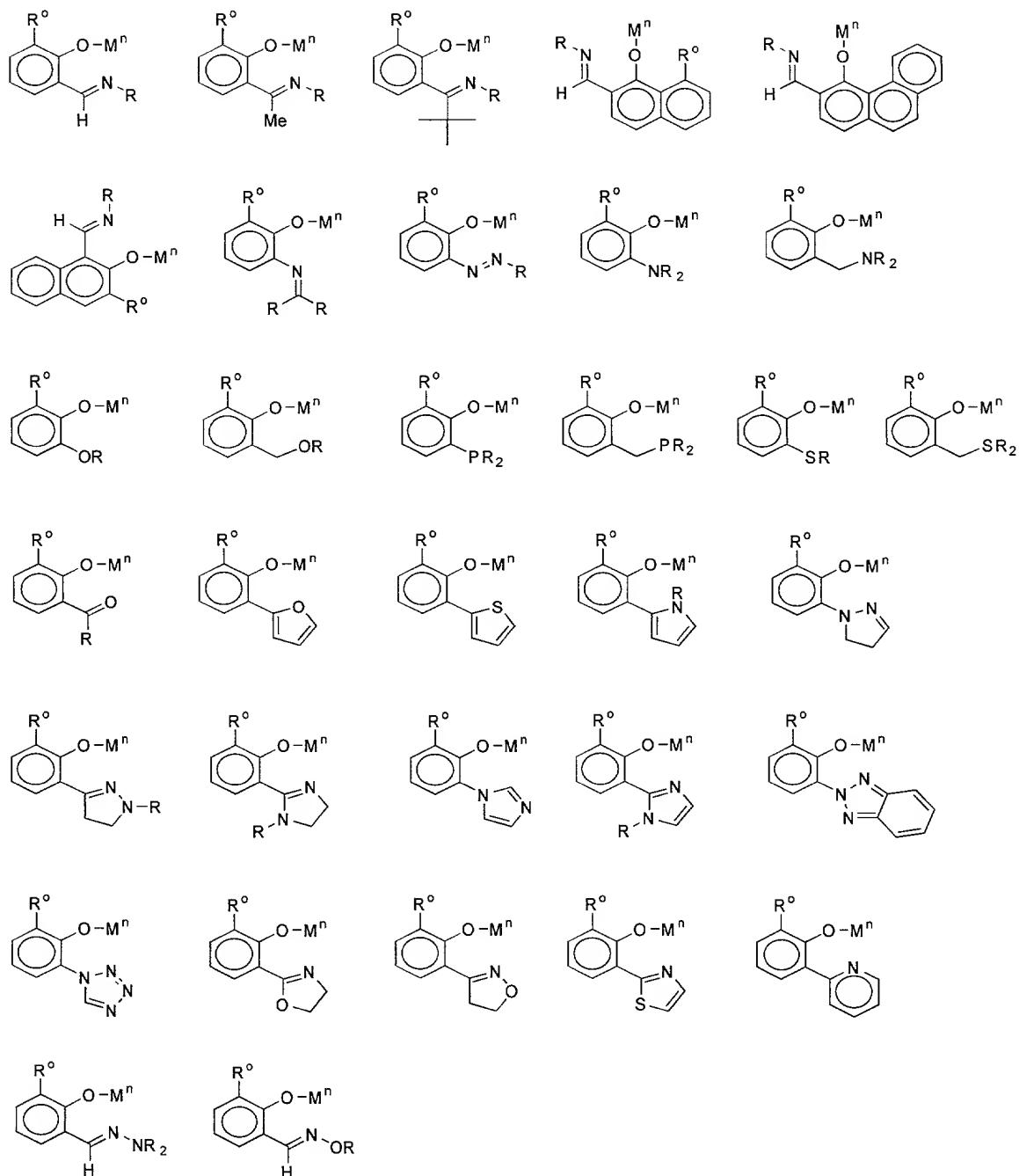


5 wherein R¹ is hydrogen or a C₄ to C₁₀₀ group, preferably a tertiary alkyl group, preferably a C₄ to C₂₀ alkyl group, preferably a C₄ to C₂₀ tertiary alkyl group, preferably a neutral C₄ to C₁₀₀ group and may or may not also be bound to M, and at least one of R² to R⁵ is a group containing a heteroatom, the rest of R² to R⁵ are independently hydrogen or a C₁ to C₁₀₀ group, preferably a C₄ to C₂₀ alkyl group (preferably butyl, isobutyl, pentyl hexyl, heptyl, isohexyl, octyl, isoctyl, decyl, nonyl, dodecyl) and any of R² to R⁵ also may or may not be bound to M, provided that in the first formula if M is a group 4 metal then the R⁴ and R⁵ groups do not form pyridine and in the second formula if M is a group 4 metal the R⁴ and R⁵ groups do not form pyridine in at least one ring, and further provided that if M is a group 4 metal then R⁵ may not be an aldehyde or an ester, and further provided that if M is nickel then R⁵ may not be an imine, O is oxygen, M is a group 4 to group 10 transition metal or lanthanide metal, preferably a group 4 metal, preferably Ti, Zr or Hf, n is the valence state of the metal M, preferably 2, 3, 4, or 5, Q is an alkyl,

halogen, benzyl, amide, carboxylate, carbamate, thiolate, hydride or alkoxide group, or a bond to an R group containing a heteroatom which may be any of R¹ to R⁵. A heteroatom containing group may be any heteroatom or a heteroatom bound to carbon silica or another heteroatom. Preferred heteroatoms include boron, aluminum, 5 silicon, nitrogen, phosphorus, arsenic, tin, lead, antimony, oxygen, selenium, tellurium. Particularly preferred heteroatoms include nitrogen, oxygen, phosphorus, and sulfur. Even more particularly preferred heteroatoms include oxygen and nitrogen. The heteroatom itself may be directly bound to the phenoxide ring or it may be bound to another atom or atoms that are bound to the phenoxide ring. The 10 heteroatom containing group may contain one or more of the same or different heteroatoms. Preferred heteroatom groups include imines, amines, oxides, phosphines, ethers, ketenes, oxoazolines heterocyclics, oxazolines, thioethers, and the like. Particularly preferred heteroatom groups include imines. Any two adjacent R groups may form a ring structure, preferably a 5 or 6 membered ring. Likewise the R 15 groups may form multi-ring structures. In one embodiment any two or more R groups do not form a 5 membered ring.

In a preferred embodiment the heteroatom substituted phenoxide transition metal compound is an iminophenoxide group 4 transition metal compound, preferably an 20 iminophenoxydezirconium compound.

Preferred catalyst systems of this invention include those comprising catalysts represented by the following formulae:



wherein

R^5 = aldimino, ketimino, alkoxy, α -alkoxymethyl, thioalkoxy, α -thioalkoxymethyl, amino, α -aminomethyl, azo, phosphino, α -phosphinomethyl, keto

or cyclic substituents such as pyrrole, furan, thiophene, imidazole, pyrazole, tetrazole, oxazoline, isoazole, thiazole.

R^o = preferably tertiary alkyl or silyl group, such as $-CMe_3$, $-CMe_2Et$, CEt_3 , $-CMe_2Ph$, $-CPh_3$, $-SiMe_3$, $-SiEt_3$, $-SiPh_3$.

5 R = is hydrogen or an alkyl, aryl, silyl group or $-OT$ where O is oxygen and T is hydrogen or an alkyl, aryl or silyl group.

M^n is a group 4 to 10 transition metal or a lanthanide metal, preferably a group 4 metal, n is the valence of M and M^n is also bound to Q_{n-1} , where Q is as defined above or any of the phenoxide groups in the above formulae.

10

The synthesis of desired ligands can be accomplished using techniques described in the literature. For example, salicylimines can be prepared by condensation of a salicylaldehyde precursor with the desired primary amine. In some instances, such as those involving less-reactive amines or aldehydes, addition of a catalytic amount of formic acid or 3 Å molecular sieves may be required. These conditions are also beneficial in the synthesis of ketimine ligands from reaction of primary amines with ortho-hydroxyketones. Phenols with heterocyclic substituents can also be prepared by standard techniques. For example, ortho-cyanophenols can be converted to oxazolines via reaction with α -aminoalcohols. Certain ligands, such as ortho-benzotriazole-substituted phenols are commercially available.

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Metallation of these acidic functionalized phenols can be accomplished by reaction with basic reagents such as $Zr(CH_2Ph)_4$, $Ti(NMe_2)_4$. Reaction of phenolic ligands with $Zr(CH_2Ph)_4$ occurs with elimination of toluene, whereas reaction with $Ti(NMe_2)_4$ proceeds via amine elimination. In both cases simple alkoxide complexes are formed, as determined by 1H NMR spectroscopy. Alternatively, ligands can be deprotonated with reagents such as $BuLi$, KH or Na metal and then reacted with metal halides, such as $ZrCl_4$ or $TiCl_4$.

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25

Preferred transition metal compounds for use in this invention include:

bis(*N*-methyl-3,5-di-*t*-butylsalicylimino)zirconium(IV) dibenzyl;

bis(*N*-ethyl-3,5-di-*t*-butylsalicylimino)zirconium(IV) dibenzyl;

bis(*N*-*iso*-propyl-3,5-di-*t*-butylsalicylimino)zirconium(IV) dibenzyl;

5 bis(*N*-*t*-butyl-3,5-di-*t*-butylsalicylimino)zirconium(IV) dibenzyl;

bis(*N*-benzyl-3,5-di-*t*-butylsalicylimino)zirconium(IV) dibenzyl;

bis(*N*-hexyl-3,5-di-*t*-butylsalicylimino)zirconium(IV) dibenzyl;

bis(*N*-phenyl-3,5-di-*t*-butylsalicylimino)zirconium(IV) dibenzyl;

bis(*N*-methyl-3,5-di-*t*-butylsalicylimino)zirconium(IV) dibenzyl;

10 bis(*N*-benzyl-3,5-di-*t*-butylsalicylimino)zirconium(IV) dichloride;

bis(*N*-benzyl-3,5-di-*t*-butylsalicylimino)zirconium(IV) dipivalate;

bis(*N*-benzyl-3,5-di-*t*-butylsalicylimino)titanium(IV) dipivalate;

bis(*N*-benzyl-3,5-di-*t*-butylsalicylimino)zirconium(IV) di(bis(dimethylamide));

bis(*N*-*iso*-propyl-3,5-di-*t*-amylsalicylimino)zirconium(IV) dibenzyl;

15 bis(*N*-*iso*-propyl-3,5-di-*t*-octylsalicylimino)zirconium(IV) dibenzyl;

bis(*N*-*iso*-propyl-3,5-di-(1',1'-dimethylbenzyl)salicylimino)zirconium(IV) dibenzyl;

bis(*N*-*iso*-propyl-3,5-di-(1',1'-dimethylbenzyl)salicylimino)titanium(IV) dibenzyl;

bis(*N*-*iso*-propyl-3,5-di-(1',1'-dimethylbenzyl)salicylimino)hafnium(IV) dibenzyl;

bis(*N*-*iso*-butyl-3,5-di-(1',1'-dimethylbenzyl)salicylimino)zirconium(IV) dibenzyl;

20 bis(*N*-*iso*-butyl-3,5-di-(1',1'-dimethylbenzyl)salicylimino)zirconium(IV) dichloride;

bis(*N*-hexyl-3,5-di-(1',1'-dimethylbenzyl)salicylimino)zirconium(IV) dibenzyl;

bis(*N*-phenyl-3,5-di-(1',1'-dimethylbenzyl)salicylimino)zirconium(IV) dibenzyl;

bis(*N*-*iso*-propyl-3,5-di-(1'-methylcyclohexyl)salicylimino)zirconium(IV) dibenzyl;

bis(*N*-benzyl-3-*t*-butylsalicylimino)zirconium(IV) dibenzyl;

25 bis(*N*-benzyl-3-triphenylmethylsalicylimino)zirconium(IV) dibenzyl;

bis(*N*-*iso*-propyl-3,5-di-trimethylsilylsalicylimino)zirconium(IV) dibenzyl;

bis(*N*-*iso*-propyl-3-(phenyl)salicylimino)zirconium(IV) dibenzyl;

bis(*N*-benzyl-3-(2',6'-di-*iso*-propylphenyl)salicylimino)zirconium(IV) dibenzyl;

bis(*N*-benzyl-3-(2',6'-di-phenylphenyl)salicylimino)zirconium(IV) dibenzyl;

bis(*N*-benzyl-3-*t*-butyl-5-methoxysalicylimino)zirconium(IV) dibenzyl;
bis(2-(2H-benzotriazol-2-yl)-4,6-di-*t*-amylphenoxy)zirconium(IV) dibenzyl;
bis(2-(2H-benzotriazol-2-yl)-4,6-di-*t*-amylphenoxy)zirconium(IV) dichloride;
bis(2-(2H-benzotriazol-2-yl)-4,6-di-*t*-amylphenoxy)zirconium(IV)
5 di(bis(dimethylamide));
bis(2-(2H-benzotriazol-2-yl)-4,6-di-(1',1'-dimethylbenzyl)phenoxy)zirconium(IV)
dibenzyl;
bis(2-(2H-benzotriazol-2-yl)-4,6-di-*t*-amylphenoxy)titanium(IV) dibenzyl;
bis(2-(2H-benzotriazol-2-yl)-4,6-di-(1',1'-dimethylbenzyl)phenoxy)titanium(IV)
10 dibenzyl;
bis(2-(2H-benzotriazol-2-yl)-4,6-di-(1',1'-dimethylbenzyl)phenoxy)titanium(IV)
dichloride;
bis(2-(2H-benzotriazol-2-yl)-4,6-di-(1',1'-dimethylbenzyl)phenoxy)hafnium(IV)
dibenzyl;
15 (*N*-phenyl-3,5-di-(1',1'-dimethylbenzyl)salicylimino)zirconium(IV) tribenzyl;
(*N*-(2',6'-di-*iso*-propylphenyl)-3,5-di-(1',1'-
dimethylbenzyl)salicylimino)zirconium(IV) tribenzyl;
(*N*-(2',6'-di-*iso*-propylphenyl)-3,5-di-(1',1'-
dimethylbenzyl)salicylimino)titanium(IV) tribenzyl; and
20 (*N*-(2',6'-di-*iso*-propylphenyl)-3,5-di-(1',1'-
dimethylbenzyl)salicylimino)zirconium(IV) trichloride.

In a preferred embodiment one or more of the transition metal compounds named
above is combined with an aluminum alkyl, an alumoxane, a modified alumoxane, a
25 non-coordinating anion, a borane or a mixture thereof.

The catalysts described herein are preferably combined with an activator to form an
olefin polymerization catalyst system. Preferred activators include alkyl aluminum
compounds (such as diethylaluminum chloride), alumoxanes, modified alumoxanes,

non-coordinating anions, boranes and the like. It is within the scope of this invention to use alumoxane or modified alumoxane as an activator, and/or to also use ionizing activators, neutral or ionic, such as tri (n-butyl) ammonium tetrakis (pentafluorophenyl) boron or a trisperfluorophenyl boron metalloid precursor which ionize the neutral metallocene compound. Boranes appear to perform better than borates, however this may be an experimental artifact and should not be construed as limiting this invention. Other useful compounds include triphenyl boron, triethyl boron, tri-n-butyl ammonium tetrethylborate, triaryl borane and the like.

There are a variety of methods for preparing alumoxane and modified alumoxanes, non-limiting examples of which are described in U.S. Patent No. 4,665,208, 4,952,540, 5,091,352, 5,206,199, 5,204,419, 4,874,734, 4,924,018, 4,908,463, 4,968,827, 5,308,815, 5,329,032, 5,248,801, 5,235,081, 5,157,137, 5,103,031, 5,391,793, 5,391,529, 5,693,838, 5,731,253 and 5,731,451 and European publications EP-A-0 561 476, EP-B1-0 279 586 and EP-A-0 594-218, and PCT publication WO 94/10180, all of which are herein fully incorporated by reference.

Ionizing compounds may contain an active proton, or some other cation associated with but not coordinated to or only loosely coordinated to the remaining ion of the ionizing compound. Such compounds and the like are described in European publications EP-A-0 570 982, EP-A-0 520 732, EP-A-0 495 375, EP-A-0 426 637, EP-A-500 944, EP-A-0 277 003 and EP-A-0 277 004, and U.S. Patent Nos. 5,153,157, 5,198,401, 5,066,741, 5,206,197, 5,241,025, 5,387,568, 5,384,299 and 5,502,124 and U.S. Patent Application Serial No. 08/285,380, filed August 3, 1994, all of which are herein fully incorporated by reference. Other activators include those described in PCT publication WO 98/07515 such as tris (2, 2', 2''-nonafluorobiphenyl) fluoroaluminate, which is fully incorporated herein by reference. Combinations of activators are also contemplated by the invention, for example, alumoxanes and ionizing activators in combinations, see for example, PCT

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publications WO 94/07928 and WO 95/14044 and U.S. Patent Nos. 5,153,157 and 5,453,410 all of which are herein fully incorporated by reference. Also, methods of activation such as using radiation and the like are also contemplated as activators for the purposes of this invention.

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In general the transition metal compound and the activator are combined in ratios of about 1000:1 to about 0.5:1. In a preferred embodiment the transition metal compound and the activator are combined in a ratio of about 300:1 to about 1:1, preferably about 10:1 to about 1:1, for boranes the ratio is preferably about 1:1 to about 10:1 and for alkyl aluminum compounds (such as diethylaluminum chloride combined with water) the ratio is preferably about 0.5:1 to about 10:1.

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In one embodiment the catalysts systems described above can further include other classes of catalysts, such as for example one or more Ziegler-Natta catalysts and/or one or more metallocene catalyst and/or one or more vanadium catalysts and/or one or more chromium catalysts. In a preferred embodiment a Ziegler-Natta catalyst as described in Ziegler-Natta Catalysts and Polymerizations, John Boor, Academic Press, New York, 1979 (with or without a separate activator) is combined with a catalyst system of this invention and used to polymerize one or more olefins. In another embodiment a metallocene catalyst (such as a cyclopentadienyl transition metal compound) with or without a separate activator is combined with a catalyst system of this invention and used to polymerize one or more olefins. Preferred cyclopentadienyl transition metal compounds are those mono-and bis-cyclopentadienyl group 4, 5 and 6 compounds described in U.S. Patent Nos. 4,530,914, 4,805,561, 4,937,299, 5,124,418, 5,017,714, 5,057,475, 5,064,802, 5,278,264, 5,278,119, 5,304,614, 5,324,800, 5,347,025, 5,350,723, 5,391,790 5,391,789, EP-A-0 591 756, EP-A-0 520 732, EP-A-0 578,838, EP-A-0 638,595, EP-A-0 420 436, WO 91/04257, WO 92/00333, WO 93/08221, WO 93/08199, WO

94/01471, WO 94/07928, WO 94/03506 and WO 95/07140, all of which are fully incorporated by reference herein.

The catalysts and catalyst systems described above can be used in any known olefin polymerization process including gas phase, solution, slurry and high pressure. The catalysts and catalyst systems described above are particularly suitable for use a solution, gas or slurry polymerization process or a combination thereof, most preferably a gas or slurry phase polymerization process.

In one embodiment, this invention is directed toward the solution, slurry or gas phase polymerization reactions involving the polymerization of one or more of monomers having from 2 to 30 carbon atoms, preferably 2-12 carbon atoms, and more preferably

2 to 8 carbon atoms. Preferred monomers include one or more of ethylene, propylene, butene-1, pentene-1, 4-methyl-pentene-1, 3,5,5,-trimethyl-hexene-1,

hexene-1, octene-1, decene-1, 3-methyl-pentene-1, and cyclic olefins or a

combination thereof. Other monomers can include vinyl monomers, diolefins such as dienes, polyenes, norbornene, norbornadiene monomers. In one embodiment, a homopolymer of ethylene is produced.

Typically in a gas phase polymerization process a continuous cycle is employed

where in one part of the cycle of a reactor system, a cycling gas stream, otherwise

known as a recycle stream or fluidizing medium, is heated in the reactor by the heat of polymerization. This heat is removed from the recycle composition in another part of the cycle by a cooling system external to the reactor. Generally, in a gas fluidized

bed process for producing polymers, a gaseous stream containing one or more

monomers is continuously cycled through a fluidized bed in the presence of a catalyst under reactive conditions. The gaseous stream is withdrawn from the fluidized bed and recycled back into the reactor. Simultaneously, polymer product is withdrawn from the reactor and fresh monomer is added to replace the polymerized monomer.

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(See for example U.S. Patent Nos. 4,543,399, 4,588,790, 5,028,670, 5,317,036, 5,352,749, 5,405,922, 5,436,304, 5,453,471, 5,462,999, 5,616,661 and 5,668,228 all of which are fully incorporated herein by reference.)

5 The reactor pressure in a gas phase process may vary from about 100 psig (690 kPa) to about 500 psig (3448 kPa), preferably in the range of from about 200 psig (1379 kPa) to about 400 psig (2759 kPa), more preferably in the range of from about 250 psig (1724 kPa) to about 350 psig (2414 kPa).

10 The reactor temperature in the gas phase process may vary from about 30°C to about 120°C, preferably from about 60°C to about 115°C, more preferably in the range of from about 70°C to 110°C, and most preferably in the range of from about 70°C to about 95°C.

15 The productivity of the catalyst or catalyst system in a gas phase system is influenced by the main monomer partial pressure. The preferred mole percent of the main monomer, ethylene or propylene, preferably ethylene, is from about 25 to 90 mole percent and the monomer partial pressure is in the range of from about 75 psia (517 kPa) to about 300 psia (2069 kPa), which are typical conditions in a gas phase polymerization process.

20

In a preferred embodiment, the reactor utilized in the present invention is capable and the process of the invention is producing greater than 500 lbs of polymer per hour (227 Kg/hr) to about 200,000 lbs/hr (90,900 Kg/hr) or higher of polymer, preferably greater than 1000 lbs/hr (455 Kg/hr), more preferably greater than 10,000 lbs/hr (4540 Kg/hr), even more preferably greater than 25,000 lbs/hr (11,300 Kg/hr), still more preferably greater than 35,000 lbs/hr (15,900 Kg/hr), still even more preferably greater than 50,000 lbs/hr (22,700 Kg/hr) and most preferably greater than 65,000 lbs/hr (29,000 Kg/hr) to greater than 100,000 lbs/hr (45,500 Kg/hr).

Other gas phase processes contemplated by the process of the invention include those described in U.S. Patent Nos. 5,627,242, 5,665,818 and 5,677,375, and European publications EP-A- 0 794 200, EP-A- 0 802 202 and EP-B- 634 421 all of which are
5 herein fully incorporated by reference.

A slurry polymerization process generally uses pressures in the range of from about 1 to about 50 atmospheres and even greater and temperatures in the range of 0°C to about 120°C. In a slurry polymerization, a suspension of solid, particulate polymer is
10 formed in a liquid polymerization diluent medium to which ethylene and comonomers along with catalyst are added. The suspension including diluent is intermittently or continuously removed from the reactor where the volatile components are separated from the polymer and recycled, optionally after a distillation, to the reactor. The liquid diluent employed in the polymerization medium is typically an alkane having from 3 to 7 carbon atoms, preferably a branched alkane.
15 The medium employed should be liquid under the conditions of polymerization and relatively inert. When a propane medium is used the process must be operated above the reaction diluent critical temperature and pressure. Preferably, a hexane or an isobutane medium is employed.
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In one embodiment, a preferred polymerization technique of the invention is referred to as a particle form polymerization, or a slurry process where the temperature is kept below the temperature at which the polymer goes into solution. Such technique is well known in the art, and described in for instance U.S. Patent No. 3,248,179 which
25 is fully incorporated herein by reference. The preferred temperature in the particle form process is within the range of about 185°F (85°C) to about 230°F (110°C). Two preferred polymerization methods for the slurry process are those employing a loop reactor and those utilizing a plurality of stirred reactors in series, parallel, or combinations thereof. Non-limiting examples of slurry processes include continuous

loop or stirred tank processes. Also, other examples of slurry processes are described in U.S. Patent No. 4,613,484, which is herein fully incorporated by reference.

In another embodiment, the slurry process is carried out continuously in a loop reactor. The catalyst as a slurry in isobutane or as a dry free flowing powder is injected regularly to the reactor loop, which is itself filled with circulating slurry of growing polymer particles in a diluent of isobutane containing monomer and comonomer. Hydrogen, optionally, may be added as a molecular weight control. The reactor is maintained at pressure of about 525 psig to 625 psig (3620 kPa to 4309 kPa) and at a temperature in the range of about 140 °F to about 220 °F (about 60 °C to about 104 °C) depending on the desired polymer density. Reaction heat is removed through the loop wall since much of the reactor is in the form of a double-jacketed pipe. The slurry is allowed to exit the reactor at regular intervals or continuously to a heated low pressure flash vessel, rotary dryer and a nitrogen purge column in sequence for removal of the isobutane diluent and all unreacted monomer and comonomers. The resulting hydrocarbon free powder is then compounded for use in various applications.

In another embodiment, the reactor used in the slurry process of the invention is capable of and the process of the invention is producing greater than 2000 lbs of polymer per hour (907 Kg/hr), more preferably greater than 5000 lbs/hr (2268 Kg/hr), and most preferably greater than 10,000 lbs/hr (4540 Kg/hr). In another embodiment the slurry reactor used in the process of the invention is producing greater than 15,000 lbs of polymer per hour (6804 Kg/hr), preferably greater than 25,000 lbs/hr (11,340 Kg/hr) to about 100,000 lbs/hr (45,500 Kg/hr).

In another embodiment in the slurry process of the invention the total reactor pressure is in the range of from 400 psig (2758 kPa) to 800 psig (5516 kPa), preferably 450 psig (3103 kPa) to about 700 psig (4827 kPa), more preferably 500 psig (3448 kPa)

to about 650 psig (4482 kPa), most preferably from about 525 psig (3620 kPa) to 625 psig (4309 kPa).

In yet another embodiment in the slurry process of the invention the concentration of 5 ethylene in the reactor liquid medium is in the range of from about 1 to 10 weight percent, preferably from about 2 to about 7 weight percent, more preferably from about 2.5 to about 6 weight percent, most preferably from about 3 to about 6 weight percent.

A preferred process of the invention is where the process, preferably a slurry or gas 10 phase process is operated in the absence of or essentially free of any scavengers, such as triethylaluminum, trimethylaluminum, tri-isobutylaluminum and tri-n-hexylaluminum and diethyl aluminum chloride, dibutyl zinc and the like. This preferred process is described in PCT publication WO 96/08520 and U.S. Patent No. 5,712,352, which are herein fully incorporated by reference.

15 In another preferred embodiment the one or all of the catalysts are tumbled with up to 6 weight % of a metal stearate, (preferably a aluminum stearate, more preferably aluminum distearate) based upon the weight of the catalyst, any support and the stearate, preferably 2 to 3 weight %. In an alternate embodiment a solution of the 20 metal stearate is fed into the reactor. These agents may be dry tumbled with the catalyst or may be fed into the reactor in a solution with or without the catalyst system or its components.

25 The catalyst and/or the activator may be placed on a support. Typically the support can be of any of the solid, porous supports. Typical support materials include talc; inorganic oxides such as silica, magnesium chloride, alumina, silica-alumina; polymeric supports such as polyethylene, polypropylene, polystyrene; and the like. Preferably the support is used in finely divided form. Prior to use the support is 30 preferably partially or completely dehydrated. The dehydration may be done physically by calcining or by chemically converting all or part of the active

hydroxyls. For more information on how to support catalysts please see US 4,808,561 which teaches how to support a metallocene catalyst system. The techniques used therein are generally applicable for this invention.

5 The catalyst system, the catalyst and or the activator may also be introduced into the reactor in solution. In one embodiment a solution of the activated catalyst in an alkane such as pentane, hexane, isopentane or the like is feed into a gas phase reactor.

10 In a preferred embodiment, the polyolefin recovered typically has a melt index as measured by ASTM D-1238, Condition E, at 190°C of 100 g/10min or less. In a preferred embodiment the polyolefin is ethylene homopolymer.

15 In a preferred embodiment the catalyst system described above is used to make a polyethylene having a density of between 0.89 and 0.960 g/cm³ (as measured by ASTM 2839), a melt index of 1.0 or less g/10min or less (as measured by ASTM D-1238, Condition E, at 190°C). Polyethylene having a melt index of between 0.01 to 10 dg/min is preferably produced. In some embodiments, a density of 0.915 to 0.940g/cm³ would be preferred, in other embodiments densities of 0.930 to 0.960g/cm³ are preferred.

20 The polyolefins then can be made into films, molded articles, sheets and the like. The films may be formed by any of the conventional technique known in the art including extrusion, co-extrusion, lamination, blowing and casting. The film may be obtained by the flat film or tubular process which may be followed by orientation in an uniaxial direction or in two mutually perpendicular directions in the plane of the film to the same or different extents. Orientation may be to the same extent in both directions or may be to different extents. Particularly preferred methods to form the polymers into films include extrusion or coextrusion on a blown or cast film line.

The films produced may further contain additives such as slip, antiblock, antioxidants, pigments, fillers, antifog, UV stabilizers, antistats, polymer processing aids, neutralizers, lubricants, surfactants, pigments, dyes and nucleating agents.

Preferred additives include silicon dioxide, synthetic silica, titanium dioxide, polydimethylsiloxane, calcium carbonate, metal stearates, calcium stearate, zinc stearate, talc, BaSO₄, diatomaceous earth, wax, carbon black, flame retarding additives, low molecular weight resins, hydrocarbon resins, glass beads and the like. The additives may be present in the typically effective amounts well known in the art, such as 0.001 weight % to 10 weight %.

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Catalyst activity and polymer molecular weight appear to vary with ligand structure. High catalyst activity was observed with ligands which contain *t*-alkyl substituents *ortho* to the phenol moiety. The 1,1-dimethylbenzyl (-CMe₂Ph) substituent had activities greater than 10⁶ g PE/mmol•Zr•hr•100 psi C₂H₄. Melt index appears to vary with the imine -NR substituent. Melt indices between 0.03 and 850 were obtained by simple variations in imine -NR. These catalysts produced broad molecular weight distribution HDPE with very small amounts of comonomer incorporation. The ability to predict melt index with changes in imine substitution and combined with outstanding activity makes these new catalysts useful for solution-feeding into gas phase polymerization reactors to obtain bimodal high density polyethylene, preferably polyethylene having a density of 0.920 to 0.950 g/cm³.

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With methyl alumoxane cocatalyst, some catalysts in this class exhibit extremely high activities for ethylene polymerization. We have observed that imine -NR substituents and salicylaldehyde substituents have an effect on catalyst activity and polymer molecular weight. Significant activities are observed with ligands which contain *t*-alkyl substituents in the position *ortho* to the phenol moiety. Activities greater than 10⁶ g PE/mmol•Zr•hr•100 psi (690kPa) C₂H₄ were observed with *ortho* -CMe₂Ph substituents. These catalysts incorporate comonomer (0.5 branches/1000 CH₂) and

produce broad molecular weight distribution HDPE. Studies indicate that, unlike metallocenes, these catalysts do not require the presence of α -olefin comonomer for high activity; similar activities are observed with and without 1-hexene.

5 This invention further relates to a library of a plurality of heteroatom substituted phenoxide group 4 to 10 transition metal or lanthanide metal compounds wherein the metal is bound to the oxygen of the phenoxide group and provided that:

- a) if more than one heteroatom substituted phenoxide is present it is not bridged to the other heteroatom substituted phenoxide,
- 10 b) if the metal is a group 4 metal, the heteroatom substituted phenoxide does not contain pyridine,
- c) if the metal is a group 4 metal then the carbon ortho to the carbon bound to the oxygen of the phenoxide may not be bound to an aldehyde or an ester, and
- 15 d) if the metal is nickel then the carbon ortho to the carbon bound to the oxygen of the phenoxide may not be bound to an imine.

In a preferred embodiment the heteroatom substituted phenoxide group 4 to 10 transition metal or lanthanide metal compounds are represented by the formulae above. These libraries may then be used for the simultaneous parallel screening of catalysts by combining the library with one or more olefins.

Examples

20 MMAO is modified methylalumoxane (type 3 in hexane) commercially available from Akzo Chemicals, Inc. under the trade name Modified Methylalumoxane type 3A, covered under patent number US 5,041,584)

Mn and Mw were measured by gel permeation chromatography on a waters 150°C GPC instrument equipped with differential refraction index detectors. The GPC

columns were calibrated by running a series of narrow polystyrene standards and the molecular weights were calculated using Mark Houwink coefficients for the polymer is question.

Density was measured according to ASTM D 1505.

5 Melt Index (MI) was measured according to ASTM D-1238, Condition E, at 190°C.

Melt Index Ratio (MIR) is the ratio of I₂₁ over I₂ as determined by ASTM D-1238.

Melt Flow Index (MFI) was measured according to ASTM 1238.

Examples

10 **General.** All manipulations were performed under nitrogen using Schlenk or drybox techniques. Toluene was dried and deoxygenated using alumina/Q-5. Hexane and C₆D₆ were distilled from Na/K. 1-Hexene was dried over alumina.

15 Slurry polymerizations were performed under 85 psi (586 kPa) (586 kPa) ethylene in a 1 L stirred reactor charged with 600 ml hexane, 43 ml hexene (unless otherwise stated) and *iso*-Bu₃Al (100 μ mol) at 75 °C. Polymerizations were performed for 30 min.

Example 1.

20 A solution of *N*-*iso*-propyl-3,5-di-*t*-butylsalicylimine (605 mg, 2.2 mmol) in *ca.* 5 ml toluene was slowly added to a solution of ZrBz₄ (500 mg, 1.1 mmol) in 50 ml toluene. The resulting dark yellow solution was stirred for 30 min. Solvent was removed *in vacuo* to yield a reddish-brown solid which was identified as bis(*N*-benzyl-3,5-di-*t*-butylsalicylimino)zirconium(IV) dibenzyl by ¹H NMR spectroscopy.

25 ¹H NMR (C₆D₆) δ 8.07 (s, HC=N, 1H), 7.77 (d, J = 2.4 Hz, saliylimine), 7.1-6.95 (m, 5H, aryl), 6.73 (t, J = 7.2 Hz, 1H, benzyl), 4.17 (septet, J = 6.6 Hz, 1H, CHMe₂), 2.76 (AB quartet, J = 10.2 Hz, 2H, ZrCH₂Ph), 1.78 (s, 9H, *t*-Bu), 1.29 (s, 9H, *t*-Bu), 0.76 (d, J = 6.6 Hz, 3H, NCHMe_AMe_B), 0.52 (d, J = 6.6 Hz, 3H, NCHMe_AMe_B).

A stock solution of bis(*N*-*iso*-propyl-3,5-di-*t*-butylsalicylimino)zirconium(IV) dibenzyl (22 mg) in 4.00 ml toluene was prepared. A 1.00 ml aliquot was removed

and added to 280 μ L of 1-hexene. MMAO (1.55 ml, 1.55 mmol, Al/Zr = 300) was slowly added over 2 minutes. After 5 additional minutes, 0.28 μ L (1 mmol Zr) of the activated catalyst solution was injected into a 1L slurry reactor, containing 600 ml hexane, 43 ml 1-hexene and 100 μ mol *iso*-Bu₃Al at 33 °C. The reactor was then 5 pressurized to 85 psi (586 kPa) with ethylene and heated to 75 °C. After 30 minutes, the reactor was cooled to ambient temperature and vented. Solid polyethylene (32.84 g, MI = 78.84) was obtained which corresponds to an activity of 77.3 kg PE/mmol Zr•hr•100 psi(690 kPa)C₂H₄. GPC analysis indicated M_W = 60,192 and M_N = 11,699 (M_W/M_N = 5.14). Infrared analysis indicated the absence of butyl branches from 1- 10 hexene incorporation.

Example 2.

Bis(*N*-benzyl-3,5-di-*t*-butylsalicylimino)zirconium(IV) dibenzyl was synthesized and activated with 300 equiv MMAO using the procedure described in 15 Example 1. Activated catalyst (1 μ mol) was injected into the polymerization reactor under reaction conditions (85 psi (586 kPa) ethylene, 75 °C). Solid polyethylene (60.96 g, MI = 0.50; FI = 49.2, MFR = 100) was obtained which corresponds to an activity of 143 kg PE/mmol Zr•hr•100 psi(690kPa)C₂H₄. GPC analysis indicated M_W = 213,460 and M_N = 15,398 (M_W/M_N = 13.86). Infrared analysis indicated the 20 absence of butyl branches from 1-hexene incorporation.

Example 3.

Bis(*N*-benzyl-3,5-di-*t*-butylsalicylimino)zirconium(IV) dibenzyl was synthesized and activated with 300 equiv MMAO using the procedure described in 25 Example 1. Activated catalyst (1 μ mol) was injected into the polymerization reactor under reaction conditions (85 psi (586 kPa) ethylene, 75 °C). No 1-hexene was added to the polymerization reactor. Solid polyethylene (23.47 g, MI = 0.06, FI = 9.95, MFR = 162) as obtained which corresponds to an activity of 55.2 kg PE/mmol Zr•hr•100 psi C₂H₄. GPC analysis indicated M_W = 167,811 and M_N = 13,609

($M_w/M_n = 12.3$). Infrared analysis indicated the absence of butyl branches from 1-hexene incorporation.

Example 4.

5 Bis(*N*-benzyl-3,5-di-*t*-butylsalicylimino)zirconium(IV) dibenzyl was synthesized and activated with 461 equiv MMAO using the procedure described in Example 1. Activated catalyst (650 nmol) was injected into the polymerization reactor at 33 °C. The concentration of 1-hexene was increased in the polymerization reactor by addition of a total of 86 ml 1-hexene. The reactor was then pressurized to 85 psi (586 10 kPa) with ethylene and heated to 75 °C. After 30 minutes, the reactor was cooled to ambient temperature and vented. Solid polyethylene (34.8 g, MI = 2.38, FI = 149, MFR = 62) as obtained which corresponds to an activity of 125.9 kg PE/mmol Zr•hr•100 psi(690kPa) C₂H₄. Infrared analysis indicated the absence of butyl branches from 1-hexene incorporation.

15 **Example 5.**

20 Bis(*N*-benzyl-3,5-di-*t*-butylsalicylimino)zirconium(IV) dibenzyl was synthesized and activated with 385 equiv MMAO using the procedure described in Example 1. Activated catalyst (650 nmol) was injected was injected into the polymerization reactor at 33 °C. In place of *iso*-Bu₃Al, an additional 385 equiv MMAO was added to the polymerization reactor. The reactor was then pressurized to 85 psi (586 kPa) with ethylene and heated to 75 °C. After 30 minutes, the reactor was cooled to ambient temperature and vented. Solid polyethylene (49.1 g, MI = 0.66, FI = 63, MFR = 96) was obtained which corresponds to an activity of 178 kg PE/mmol Zr•hr•100 psi(690kPa) C₂H₄. Infrared analysis indicated the absence of butyl branches from 1-hexene incorporation.

25 **Example 6.**

Bis(*N*-benzyl-3,5-di-*t*-butylsalicylimino)zirconium(IV) dibenzyl was synthesized and activated with 461 equiv MMAO using the procedure described in Example 1. Activated catalyst (650 nmol) was injected into the polymerization reactor at 33 °C. The reactor was then pressurized to 85 psi (586 kPa) with ethylene and heated to 75 °C. Hydrogen (2.5 psi) was added to the reactor. After 30 minutes, the reactor was cooled to ambient temperature and vented. Solid polyethylene (32.9 g, MI = 15.8; FI = 706, MFR = 45) was obtained which corresponds to an activity of 119 kg PE/mmol Zr•hr•100 psi(690kPa) C₂H₄. Infrared analysis indicated the absence of butyl branches from 1-hexene incorporation.

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Example 7.

Bis(*N*-benzyl-3,5-di-*t*-butylsalicylimino)zirconium(IV) dibenzyl was synthesized and activated with 461 equiv MAO using the procedure described in Example 1. Activated catalyst (650 nmol) was injected into the polymerization reactor at 33 °C. The reactor was then pressurized to 85 psi (586 kPa) with ethylene and heated to 75 °C. After 30 minutes, the reactor was cooled to ambient temperature and vented. Solid polyethylene (19.9 g, MI = 0.33; FI = 32.2, MFR = 96) was obtained which corresponds to an activity of 72.2 kg PE/mmol Zr•hr•100 psi(690kPa) C₂H₄. Infrared analysis indicated the absence of butyl branches from 1-hexene incorporation.

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Example 8.

Bis(*N*-benzyl-3,5-di-*t*-butylsalicylimino)zirconium(IV) dibenzyl was synthesized and activated with 154 equiv MMAO using the procedure described in Example 1. Activated catalyst (650 nmol) was injected into the polymerization reactor at 33 °C. The reactor was then pressurized to 85 psi (586 kPa) with ethylene and heated to 75 °C. After 30 minutes, the reactor was cooled to ambient temperature and vented. Solid polyethylene (35.5 g, MI = 0.85; FI = 67.3, MFR = 79) was obtained which corresponds to an activity of 128 kg PE/mmol Zr•hr•100

psi(690kPa)C₂H₄. Infrared analysis indicated the absence of butyl branches from 1-hexene incorporation.

Example 9.

5 Bis(*N*-benzyl-3,5-di-*t*-butylsalicylimino)zirconium(IV) dibenzyl was synthesized and activated with 77 equiv MMAO using the procedure described in Example 1. Activated catalyst (650 nmol) was injected into the polymerization reactor at 33 °C. The reactor was then pressurized to 85 psi (586 kPa) with ethylene and heated to 75 °C. After 30 minutes, the reactor was cooled to ambient temperature 10 and vented. Solid polyethylene (33.6 g, MI = 0.43; FI = 37.5, MFR = 88) was obtained which corresponds to an activity of 122 kg PE/mmol Zr•hr•100 psi(690kPa) C₂H₄. Infrared analysis indicated the absence of butyl branches from 1-hexene incorporation.

15 **Example 10.**

16 Bis(*N*-*iso*-butyl-3,5-di-*t*-butylsalicylimino)zirconium(IV) dibenzyl was synthesized and activated with 300 equiv MMAO using the procedure described in Example 1. Activated catalyst (500 nmol) was injected into the polymerization reactor at 33 °C. The reactor was then pressurized to 85 psi (586 kPa) with ethylene and heated to 75 °C. After 30 minutes, the reactor was cooled to ambient temperature 20 and vented. Solid polyethylene (60.42 g) was obtained which corresponds to an activity of 284 kg PE/mmol Zr•hr•100 psi C₂H₄. GPC analysis indicated M_w = 30,800 and M_N = 6110 (M_w/M_N = 5.04). Infrared analysis indicated the absence of butyl branches from 1-hexene incorporation.

25 **Example 11.**

26 Bis(*N*-*n*-hexyl-3,5-di-*t*-butylsalicylimino)zirconium(IV) dibenzyl was synthesized and activated with 300 equiv MMAO using the procedure described in Example 1. Activated catalyst (1.00 μmol) was injected into the polymerization

reactor at 33 °C. The reactor was then pressurized to 85 psi (586 kPa) with ethylene and heated to 75 °C. After 30 minutes, the reactor was cooled to ambient temperature and vented. Solid polyethylene (91.71 g) was obtained which corresponds to an activity of 216 kg PE/mmol Zr•hr•100 psi (690kPa)C₂H₄. Infrared analysis indicated the absence of butyl branches from 1-hexene incorporation.

5 **Example 12.**

10 Bis(*N*-methyl-3,5-di-*t*-butylsalicylimino)zirconium(IV) dibenzyl was synthesized and activated with 300 equiv MMAO using the procedure described in Example 1. Activated catalyst (1.00 μmol) was injected into the polymerization reactor at 33 °C. The reactor was then pressurized to 85 psi (586 kPa) with ethylene and heated to 75 °C. After 30 minutes, the reactor was cooled to ambient temperature and vented. Solid polyethylene (24.0 g, MI = 856) was obtained which corresponds to an activity of 56.5 kg PE/mmol Zr•hr•100 psi (690kPa) C₂H₄. Infrared analysis indicated the absence of butyl branches from 1-hexene incorporation.

15 **Example 13.**

20 Bis(*N*-phenyl-3,5-di-*t*-butylsalicylimino)zirconium(IV) dibenzyl was synthesized and activated with 300 equiv MMAO using the procedure described in Example 1. Activated catalyst (5.00 μmol) was injected into the polymerization reactor at ## °C. The reactor was then pressurized to 85 psi (586 kPa) with ethylene and heated to 75 °C. After 30 minutes, the reactor was cooled to ambient temperature and vented. Solid polyethylene (44.65 g, MI = 0.03, FI = 2.90, MFR = 110) was obtained which corresponds to an activity of 21.0 kg PE/mmol Zr•hr•100 psi (690kPa) C₂H₄. Infrared analysis indicated the absence of butyl branches from 1-hexene incorporation.

25 **Example 14.**

Bis(*N*-*iso*-propyl-3-*t*-butylsalicylimino)zirconium(IV) dibenzyl was synthesized and activated with 300 equiv MMAO using the procedure described in Example 1. Activated catalyst (250 nmol) was injected into the polymerization reactor at 33 °C. The reactor was then pressurized to 85 psi (586 kPa) with ethylene and heated to 75 °C. After 30 minutes, the reactor was cooled to ambient temperature and vented. Solid polyethylene (37.9 g) was obtained which corresponds to an activity of 357 kg PE/mmol Zr•hr•100 psi (690kPa) C₂H₄. Infrared analysis indicated the absence of butyl branches from 1-hexene incorporation.

Example 15.

Bis(*N*-*iso*-propyl-3,5-di-*t*-amylsalicylimino)zirconium(IV) dibenzyl was synthesized and activated with 300 equiv MMAO using the procedure described in Example 1. Activated catalyst (125 nmol) was injected into the polymerization reactor at 33 °C. The reactor was then pressurized to 85 psi (586 kPa) with ethylene and heated to 75 °C. After 30 minutes, the reactor was cooled to ambient temperature and vented. Solid polyethylene (10.34 g, 188 MI) was obtained which corresponds to an activity of 195 kg PE/mmol Zr•hr•100 psi (690kPa) C₂H₄. Infrared analysis indicated the absence of butyl branches from 1-hexene incorporation.

Example 16.

Bis(*N*-*iso*-propyl-3-adamantyl-5-methylsalicylimino)zirconium(IV) dibenzyl was synthesized and activated with 300 equiv MMAO using the procedure described in Example 1. Activated catalyst (250 nmol) was injected into the polymerization reactor at 33 °C. The reactor was then pressurized to 85 psi (586 kPa) with ethylene and heated to 75 °C. After 30 minutes, the reactor was cooled to ambient temperature and vented. Solid polyethylene (21.80 g, 76.8 MI) was obtained which corresponds to an activity of 205 kg PE/mmol Zr•hr•100 psi (690kPa) C₂H₄. Infrared analysis indicated the absence of butyl branches from 1-hexene incorporation.

Example 17.

Bis(*N*-*iso*-propyl-3-*t*-butyl-5-methoxysalicylimino)zirconium(IV) dibenzyl was synthesized and activated with 300 equiv MMAO using the procedure described in Example 1. Activated catalyst (125 nmol) was injected into the polymerization reactor at 33 °C. The reactor was then pressurized to 85 psi (586 kPa) with ethylene and heated to 75 °C. After 30 minutes, the reactor was cooled to ambient temperature and vented. Solid polyethylene (20.8 g, 186 MI) was obtained which corresponds to an activity of 391 kg PE/mmol Zr•hr•100 psi (690kPa) C₂H₄. Infrared analysis indicated the absence of butyl branches from 1-hexene incorporation.

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Example 18.

Bis(*N*-*iso*-propyl-3,5-di-(1',1'-dimethylbenzyl)salicylimino)zirconium(IV) dibenzyl was synthesized and activated with 300 equiv MMAO using the procedure described in Example 1. Activated catalyst (125 nmol) was injected into the polymerization reactor at 33 °C. The reactor was then pressurized to 85 psi (586 kPa) with ethylene and heated to 75 °C. After 30 minutes, the reactor was cooled to ambient temperature and vented. Solid polyethylene (75.65 g) was obtained which corresponds to an activity of 1,420 kg PE/mmol Zr•hr•100 psi (690kPa) C₂H₄. GPC analysis indicated M_W = 21,365 and M_N = 8355 (M_W/M_N = 2.56). Infrared analysis indicated the absence of butyl branches from 1-hexene incorporation.

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Example 19.

Bis(*N*-*iso*-propyl-3,5-di-(1',1'-dimethylbenzyl)salicylimino)zirconium(IV) dibenzyl was synthesized and activated with 25 equiv MMAO using the procedure described in Example 1. Activated catalyst (250 nmol) was injected into the polymerization reactor at 33 °C. The reactor was then pressurized to 85 psi (586 kPa) with ethylene and heated to 75 °C. After 30 minutes, the reactor was cooled to ambient temperature and vented. Solid polyethylene (81.4 g) was obtained which

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corresponds to an activity of 766 kg PE/mmol Zr•hr•100 psi (690kPa) C₂H₄. Infrared analysis indicated the absence of butyl branches from 1-hexene incorporation.

Example 20.

Bis(*N*-*iso*-propyl-3,5-di-(1',1'-dimethylbenzyl)salicylimino)zirconium(IV) dibenzyl was synthesized and activated with 10 equiv MMAO using the procedure described in Example 1. Activated catalyst (250 nmol) was injected into the polymerization reactor at 33 °C. The reactor was then pressurized to 85 psi (586 kPa) with ethylene and heated to 75 °C. After 30 minutes, the reactor was cooled to ambient temperature and vented. Solid polyethylene (8.24 g) was obtained which corresponds to an activity of 77.5 kg PE/mmol Zr•hr•100 psi (690kPa) C₂H₄. Infrared analysis indicated the absence of butyl branches from 1-hexene incorporation.

Example 21.

Bis(*N*-*iso*-propyl-3,5-di-(1',1'-dimethylbenzyl)salicylimino)zirconium(IV) dibenzyl was synthesized and activated with 300 equiv IBAO (*iso*-butyl aluminoxane, hydrolysis ration = 0.85) using the procedure described in Example 1. Activated catalyst (1.00 µmol) was injected into the polymerization reactor at 33 °C. The reactor was then pressurized to 85 psi (586 kPa) with ethylene and heated to 75 °C. After 30 minutes, the reactor was cooled to ambient temperature and vented. Solid polyethylene (75.27 g) was obtained which corresponds to an activity of 177 kg PE/mmol Zr•hr•100 psi (690kPa) C₂H₄. Infrared analysis indicated the absence of butyl branches from 1-hexene incorporation.

Example 22.

Bis(*N*-*iso*-propyl-3,5-di-(1',1'-dimethylbenzyl)salicylimino)zirconium(IV) dibenzyl was synthesized and activated with 1.0 equiv Et₂AlCl using the procedure described in Example 1. Activated catalyst (1.00 µmol) was injected into the polymerization reactor at 33 °C. The reactor was then pressurized to 85 psi (586 kPa)

with ethylene and heated to 75 °C. After 30 minutes, the reactor was cooled to ambient temperature and vented. Solid polyethylene (30.1 g) was obtained which corresponds to an activity of 70.8 kg PE/mmol Zr•hr•100 psi (690kPa) C₂H₄. Infrared analysis indicated the absence of butyl branches from 1-hexene incorporation.

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Example 23.

Bis(*N*-*iso*-propyl-3,5-di-(1',1'-dimethylbenzyl)salicylimino)zirconium(IV) dibenzyl was synthesized and activated with 2.0 equiv (C₆F₅)₃B using the procedure described in Example 1. Activated catalyst (1.00 µmol) was injected into the polymerization reactor at 33 °C. The reactor was then pressurized to 85 psi (586 kPa) with ethylene and heated to 75 °C. After 30 minutes, the reactor was cooled to ambient temperature and vented. Solid polyethylene (6.63 g) was obtained which corresponds to an activity of 15.6 kg PE/mmol Zr•hr•100 psi (690kPa) C₂H₄. Infrared analysis indicated the absence of butyl branches from 1-hexene incorporation.

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Example 24.

Bis(*N*-*iso*-propyl-3,5-di-(1',1'-dimethylbenzyl)salicylimino)zirconium(IV) dibenzyl was synthesized and activated with 2.0 equiv Ph₃B using the procedure described in Example 1. Activated catalyst (2.00 µmol) was injected into the polymerization reactor at 33 °C. The reactor was then pressurized to 85 psi (586 kPa) with ethylene and heated to 75 °C. After 30 minutes, the reactor was cooled to ambient temperature and vented. Solid polyethylene (10.4 g) was obtained which corresponds to an activity of 12.2 kg PE/mmol Zr•hr•100 psi (690kPa) C₂H₄. Infrared analysis indicated the absence of butyl branches from 1-hexene incorporation.

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Example 25.

Bis(*N*-hexyl-3,5-di-(1',1'-dimethylbenzyl)salicylimino)zirconium(IV) dibenzyl was synthesized and activated with 300 equiv MMAO using the procedure described in Example 1. Activated catalyst (250 nmol) was injected into the

polymerization reactor at 33 °C. The reactor was then pressurized to 85 psi (586 kPa) with ethylene and heated to 75 °C. After 30 minutes, the reactor was cooled to ambient temperature and vented. Solid polyethylene (95.54 g) was obtained which corresponds to an activity of 899 kg PE/mmol Zr•hr•100 psi (690kPa) C₂H₄. GPC analysis indicated M_W = 11,627 and M_N = 6054 (M_W/M_N = 2.30). Infrared analysis indicated the absence of butyl branches from 1-hexene incorporation.

Example 26.

Bis(*N*-*iso*-butyl-3,5-di-(1',1'-dimethylbenzyl)salicylimino)zirconium(IV) dibenzyl was synthesized and activated with 300 equiv MMAO using the procedure described in Example 1. Activated catalyst (250 nmol) was injected into the polymerization reactor at 33 °C. The reactor was then pressurized to 85 psi (586 kPa) with ethylene and heated to 75 °C. After 30 minutes, the reactor was cooled to ambient temperature and vented. Solid polyethylene (81.69 g) was obtained which corresponds to an activity of 769 kg PE/mmol Zr•hr•100 psi (690kPa) C₂H₄. GPC analysis indicated M_W = 11,627 and M_N = 6054 (M_W/M_N = 2.30). Infrared analysis indicated the absence of butyl branches from 1-hexene incorporation.

Example 27.

Bis(*N*-methyl-3,5-di-(1',1'-dimethylbenzyl)salicylimino)zirconium(IV) dibenzyl was synthesized and activated with 300 equiv MMAO using the procedure described in Example 1. Activated catalyst (250 nmol) was injected into the polymerization reactor at 33 °C. The reactor was then pressurized to 85 psi (586 kPa) with ethylene and heated to 75 °C. After 30 minutes, the reactor was cooled to ambient temperature and vented. Solid polyethylene (28.24 g) was obtained which corresponds to an activity of 266 kg PE/mmol Zr•hr•100 psi (690kPa) C₂H₄. Infrared analysis indicated the absence of butyl branches from 1-hexene incorporation.

Example 28.

Bis(*N*-benzyl-3,5-di-(1',1'-dimethylbenzyl)salicylimino)zirconium(IV) dibenzyl was synthesized and activated with 300 equiv MMAO using the procedure described in Example 1. Activated catalyst (250 nmol) was injected into the polymerization reactor at 33 °C. The reactor was then pressurized to 85 psi (586 kPa) with ethylene and heated to 75 °C. After 30 minutes, the reactor was cooled to ambient temperature and vented. Solid polyethylene (85.95 g, MI = 0.46, FI = 35.2, MFR = 75.9) was obtained which corresponds to an activity of 809 kg PE/mmol Zr·hr·100 psi (690kPa) C₂H₄. Infrared analysis indicated the absence of butyl branches from 1-hexene incorporation.

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The catalysts described herein were all found to produce HDPE under ethylene-hexene copolymerization conditions. FTIR measurements did not detect the presence of butyl branches. HDPE produced with these catalysts had large MFR values due to broad molecular weight distribution.

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All documents described herein are incorporated by reference herein, including any priority documents and/or testing procedures. As is apparent from the foregoing general description and the specific embodiments, while forms of the invention have been illustrated and described, various modifications can be made without departing from the spirit and scope of the invention. Accordingly it is not intended that the invention be limited thereby.

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Claims

1. A catalyst system comprising an activator and one or more heteroatom substituted phenoxide group 4 to 10 transition or lanthanide metal compounds wherein the metal is bound to the oxygen of the phenoxide group and provided that:

5 a) if more than one heteroatom substituted phenoxide is present it is not bridged to the other heteroatom substituted phenoxide,

10 b) if the metal is a group 4 metal, the heteroatom substituted phenoxide does not contain pyridine,

15 c) if the metal is a group 4 metal then the carbon ortho to the carbon bound to the oxygen of the phenoxide may not be bound to an aldehyde or an ester, and

20 d) if the metal is nickel then the carbon ortho to the carbon bound to the oxygen of the phenoxide may not be bound to an imine.

15 2. The catalyst system of claim 1 wherein the activator is an aluminum alkyl, an alumoxane, a modified alumoxane, a borane or a non-coordinating anion.

20 3. The catalyst system of claim 1 wherein the transition metal is a group 4 metal.

25 4. The catalyst system of claim 1 wherein the transition metal is zirconium.

5. The catalyst system of claim 1 wherein the heteroatom substituted phenoxide transition metal compound is an iminophenoxide group 4 transition metal compound.

25 6. The catalyst system of claim 1 wherein the heteroatom substituted phenoxide transition metal compound is an iminophenoxidezirconium compound.

7. The catalyst system of claim 1 wherein the heteroatom substituted phenoxide transition metal compound is selected from the group consisting of:

bis(*N*-methyl-3,5-di-*t*-butylsalicylimino)zirconium(IV) dibenzyl;
 bis(*N*-ethyl-3,5-di-*t*-butylsalicylimino)zirconium(IV) dibenzyl;
 bis(*N*-*iso*-propyl-3,5-di-*t*-butylsalicylimino)zirconium(IV) dibenzyl;
 bis(*N*-*t*-butyl-3,5-di-*t*-butylsalicylimino)zirconium(IV) dibenzyl;
 5 bis(*N*-benzyl-3,5-di-*t*-butylsalicylimino)zirconium(IV) dibenzyl;
 bis(*N*-hexyl-3,5-di-*t*-butylsalicylimino)zirconium(IV) dibenzyl;
 bis(*N*-phenyl-3,5-di-*t*-butylsalicylimino)zirconium(IV) dibenzyl;
 bis(*N*-methyl-3,5-di-*t*-butylsalicylimino)zirconium(IV) dibenzyl;
 10 bis(*N*-benzyl-3,5-di-*t*-butylsalicylimino)zirconium(IV) dichloride;
 bis(*N*-benzyl-3,5-di-*t*-butylsalicylimino)zirconium(IV) dipivalate;
 bis(*N*-benzyl-3,5-di-*t*-butylsalicylimino)titanium(IV) dipivalate;
 bis(*N*-benzyl-3,5-di-*t*-butylsalicylimino)zirconium(IV) di(bis(dimethylamide));
 bis(*N*-*iso*-propyl-3,5-di-*t*-amylsalicylimino)zirconium(IV) dibenzyl;
 15 bis(*N*-*iso*-propyl-3,5-di-*t*-octylsalicylimino)zirconium(IV) dibenzyl;
 bis(*N*-*iso*-propyl-3,5-di-(1',1'-dimethylbenzyl)salicylimino)zirconium(IV) dibenzyl;
 bis(*N*-*iso*-propyl-3,5-di-(1',1'-dimethylbenzyl)salicylimino)titanium(IV) dibenzyl;
 bis(*N*-*iso*-propyl-3,5-di-(1',1'-dimethylbenzyl)salicylimino)hafnium(IV) dibenzyl;
 20 bis(*N*-*iso*-butyl-3,5-di-(1',1'-dimethylbenzyl)salicylimino)zirconium(IV) dibenzyl;
 bis(*N*-*iso*-butyl-3,5-di-(1',1'-dimethylbenzyl)salicylimino)zirconium(IV) dichloride;
 bis(*N*-hexyl-3,5-di-(1',1'-dimethylbenzyl)salicylimino)zirconium(IV) dibenzyl;
 bis(*N*-phenyl-3,5-di-(1',1'-dimethylbenzyl)salicylimino)zirconium(IV) dibenzyl;
 bis(*N*-*iso*-propyl-3,5-di-(1'-methylcyclohexyl)salicylimino)zirconium(IV) dibenzyl;
 bis(*N*-benzyl-3-*t*-butylsalicylimino)zirconium(IV) dibenzyl;
 25 bis(*N*-benzyl-3-triphenylmethylsalicylimino)zirconium(IV) dibenzyl;
 bis(*N*-*iso*-propyl-3,5-di-trimethylsilylsalicylimino)zirconium(IV) dibenzyl;
 bis(*N*-*iso*-propyl-3-(phenyl)salicylimino)zirconium(IV) dibenzyl;
 bis(*N*-benzyl-3-(2',6'-di-*iso*-propylphenyl)salicylimino)zirconium(IV) dibenzyl;
 bis(*N*-benzyl-3-(2',6'-di-phenylphenyl)salicylimino)zirconium(IV) dibenzyl;
 bis(*N*-benzyl-3-*t*-butyl-5-methoxysalicylimino)zirconium(IV) dibenzyl;

bis(2-(2H-benzotriazol-2-yl)-4,6-di-*t*-amylphenoxyde)zirconium(IV) dibenzyl;
bis(2-(2H-benzotriazol-2-yl)-4,6-di-*t*-amylphenoxyde)zirconium(IV) dichloride;
bis(2-(2H-benzotriazol-2-yl)-4,6-di-*t*-amylphenoxyde)zirconium(IV)
di(bis(dimethylamide));
5 bis(2-(2H-benzotriazol-2-yl)-4,6-di-(1',1'-dimethylbenzyl)phenoxyde)zirconium(IV)
dibenzyl;
bis(2-(2H-benzotriazol-2-yl)-4,6-di-*t*-amylphenoxyde)titanium(IV) dibenzyl;
bis(2-(2H-benzotriazol-2-yl)-4,6-di-(1',1'-dimethylbenzyl)phenoxyde)titanium(IV)
dibenzyl;
10 bis(2-(2H-benzotriazol-2-yl)-4,6-di-(1',1'-dimethylbenzyl)phenoxyde)titanium(IV)
dichloride;
bis(2-(2H-benzotriazol-2-yl)-4,6-di-(1',1'-dimethylbenzyl)phenoxyde)hafnium(IV)
dibenzyl;
15 (N-phenyl-3,5-di-(1',1'-dimethylbenzyl)salicylimino)zirconium(IV) tribenzyl
(N-(2',6'-di-*iso*-propylphenyl)-3,5-di-(1',1'-
dimethylbenzyl)salicylimino)zirconium(IV) tribenzyl;
(N-(2',6'-di-*iso*-propylphenyl)-3,5-di-(1',1'-
dimethylbenzyl)salicylimino)titanium(IV) tribenzyl; and
20 (N-(2',6'-di-*iso*-propylphenyl)-3,5-di-(1',1'-
dimethylbenzyl)salicylimino)zirconium(IV) trichloride.

8. The catalyst system of claim 6 further comprising an activator comprising one or more of an aluminum alkyl, an alumoxane, a modified alumoxane, a borane or a non-coordinating anion.

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9. The catalyst system of claim 1 wherein either the transition metal compound or the activator or both are placed on a support.

10. The catalyst system of claim 1 further comprising a Ziegler-Natta catalyst.

11. The catalyst system of claim 1 further comprising a mono- or bis-cyclopentadienyl group 4, 5 and 6 transition metal compound and an optional second activator.

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12. The catalyst system of claim 1 further comprising a second a heteroatom substituted phenoxide group 4 to 10 transition metal compound and an optional second activator.

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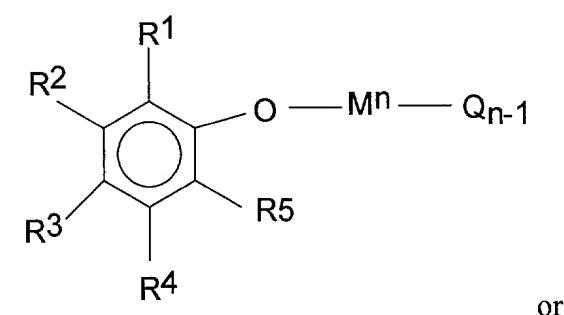
13. The catalyst system of claim 1 further comprising a second activator.

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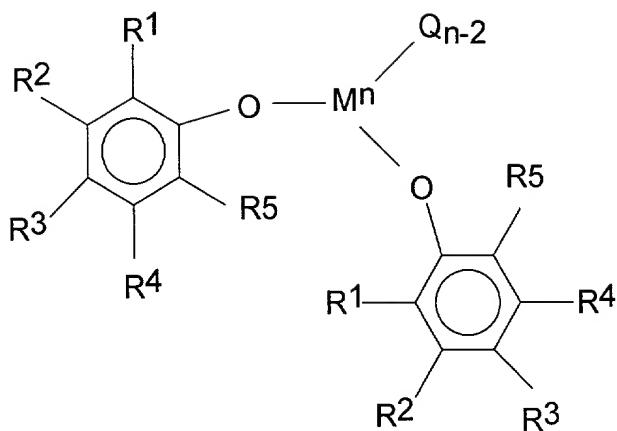
14. The catalyst system of claim 7 wherein the activator is one or more of alumoxane, tris (2, 2', 2"- nonafluorobiphenyl) fluoroaluminate, triphenyl boron, triethyl boron, tri-n-butyl ammonium tetraethylborate, triaryl borane, tri (n-butyl) ammonium tetrakis (pentafluorophenyl) boron or a trisperfluorophenyl boron, or diethylaluminum chloride.

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15. A catalyst system comprising the reaction product of an activator and one or more heteroatom substituted phenoxide transition metal compounds represented by the following formulae:



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wherein:

5 R^1 to R^5 may be independently hydrogen, a heteroatom containing group or a C_1 to C_{100} group provided that at least one of R^2 to R^5 is a group containing a heteroatom, any of R^1 to R^5 may or may not be bound to the metal M,

10 O is oxygen,

 M is a group 4 to 10 transition metal or a lanthanide metal,

 n is the valence state of M,

15 Q is an anionic ligand or a bond to an R group containing a heteroatom which may be any of R^1 to R^5 , and

 further provided that:

a) if M is a group 4 metal then R^5 may not be an aldehyde or an ester;

b) if M is nickel then R^5 may not be an imine;

15 c) the R^4 and R^5 groups do not form pyridine in the first formula if M is a group 4 metal; and

d) the R^4 and R^5 groups do not form pyridine in at least one ring of the second formula if M is a group 4 metal.

20 16. The catalyst system of claim 15 wherein the activator is an aluminum alkyl, an alumoxane, a modified alumoxane, a borane, a non-coordinating anion or a mixture thereof.

17. The catalyst system of claim 15 wherein Q is a bond to any of R² to R⁵ and the R group that Q is bound to is a heteroatom containing group.

5 18. The catalyst system of claim 15 wherein the heteroatom containing group is an imine, triazole, or oxazole.

19. The catalyst system of claim 15 wherein the heteroatom in the heteroatom containing group is nitrogen and/or oxygen.

10 20. The catalyst system of claim 15 wherein the R¹ group is a C₄ to C₂₀ alkyl group.

15 21. The catalyst system of claim 15 wherein R¹ is a tertiary alkyl group.

22. The catalyst system of claim 15 wherein R⁵ is bound to the metal.

15 23. The catalyst system of claim 15 wherein the R² group is a butyl, isobutyl, tertiary butyl, pentyl hexyl, heptyl, isohexyl, octyl, isoocetyl, decyl, nonyl, or dodecyl group.

20 24. The catalyst system of claim 15 wherein two or more R groups have formed a five or six membered ring.

25 25. The catalyst system of claim 15 wherein two or more R groups have formed a multi-ring system.

26. The catalyst system of claim 15 wherein M is zirconium, titanium or hafnium.

27. The catalyst system of claim 15 wherein n is 4.

28. The catalyst system of claim 15 wherein n is 3.

5 29. The catalyst system of claim 15 wherein Q is a halogen or an alkyl group.

30. The catalyst system of claim 15 wherein Q is an amide, carboxylate, carbamate, thiolate, hydride or alkoxide group.

10 31. The catalyst system of claim 15 further comprising a support.

32. The catalyst system of claim 15 wherein either the transition metal compound or the activator or the reaction product thereof are placed on a support selected from the group consisting of talc; silica, magnesium chloride, alumina, silica-alumina; polyethylene, polypropylene, polystyrene; or a mixture thereof.

15 33. The catalyst system of claim 32 wherein prior to being combined with the transition metal compound and/or the activator and/or the reaction product thereof the support is partially or completely dehydrated.

20 34. The catalyst system of claim 15 wherein the transition metal compound and the activator are combined in ratios of about 1000:1 to about 0.5:1.

35. The catalyst system of claim 15 wherein the transition metal compound and the activator are combined in ratios of about 300:1 to about 1:1.

25 36. The catalyst system of claim 15 wherein the activator is a borane and the transition metal compound and the borane are combined in ratios of about 1:1 to about 10:1

37. The catalyst system of claim 15 wherein the activator is an alkyl aluminum compound and the transition metal compound and the alkyl aluminum compound are combined in ratios of about 0.5:1 to about 10:1

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38. The catalyst system of claim 15 wherein two or more R groups do not form a five membered ring.

39. The catalyst system of claim 15 wherein M is zirconium.

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40. A catalyst system comprising the reaction product of an alumoxane and a one or more heteroatom substituted phenoxide transition metal compounds selected from the group consisting of:

bis(*N*-methyl-3,5-di-*t*-butylsalicylimino)zirconium(IV) dibenzyl;
bis(*N*-ethyl-3,5-di-*t*-butylsalicylimino)zirconium(IV) dibenzyl;
bis(*N*-*iso*-propyl-3,5-di-*t*-butylsalicylimino)zirconium(IV) dibenzyl;
bis(*N*-*t*-butyl-3,5-di-*t*-butylsalicylimino)zirconium(IV) dibenzyl;
bis(*N*-benzyl-3,5-di-*t*-butylsalicylimino)zirconium(IV) dibenzyl;
bis(*N*-hexyl-3,5-di-*t*-butylsalicylimino)zirconium(IV) dibenzyl;
bis(*N*-phenyl-3,5-di-*t*-butylsalicylimino)zirconium(IV) dibenzyl;
bis(*N*-methyl-3,5-di-*t*-butylsalicylimino)zirconium(IV) dibenzyl;
bis(*N*-benzyl-3,5-di-*t*-butylsalicylimino)zirconium(IV) dichloride;
bis(*N*-benzyl-3,5-di-*t*-butylsalicylimino)zirconium(IV) dipivalate;
bis(*N*-benzyl-3,5-di-*t*-butylsalicylimino)titanium(IV) dipivalate;
bis(*N*-benzyl-3,5-di-*t*-butylsalicylimino)zirconium(IV) di(bis(dimethylamide));
bis(*N*-*iso*-propyl-3,5-di-*t*-amylsalicylimino)zirconium(IV) dibenzyl;
bis(*N*-*iso*-propyl-3,5-di-*t*-octylsalicylimino)zirconium(IV) dibenzyl;
bis(*N*-*iso*-propyl-3,5-di-(1',1'-dimethylbenzyl)salicylimino)zirconium(IV) dibenzyl;
bis(*N*-*iso*-propyl-3,5-di-(1',1'-dimethylbenzyl)salicylimino)titanium(IV) dibenzyl;

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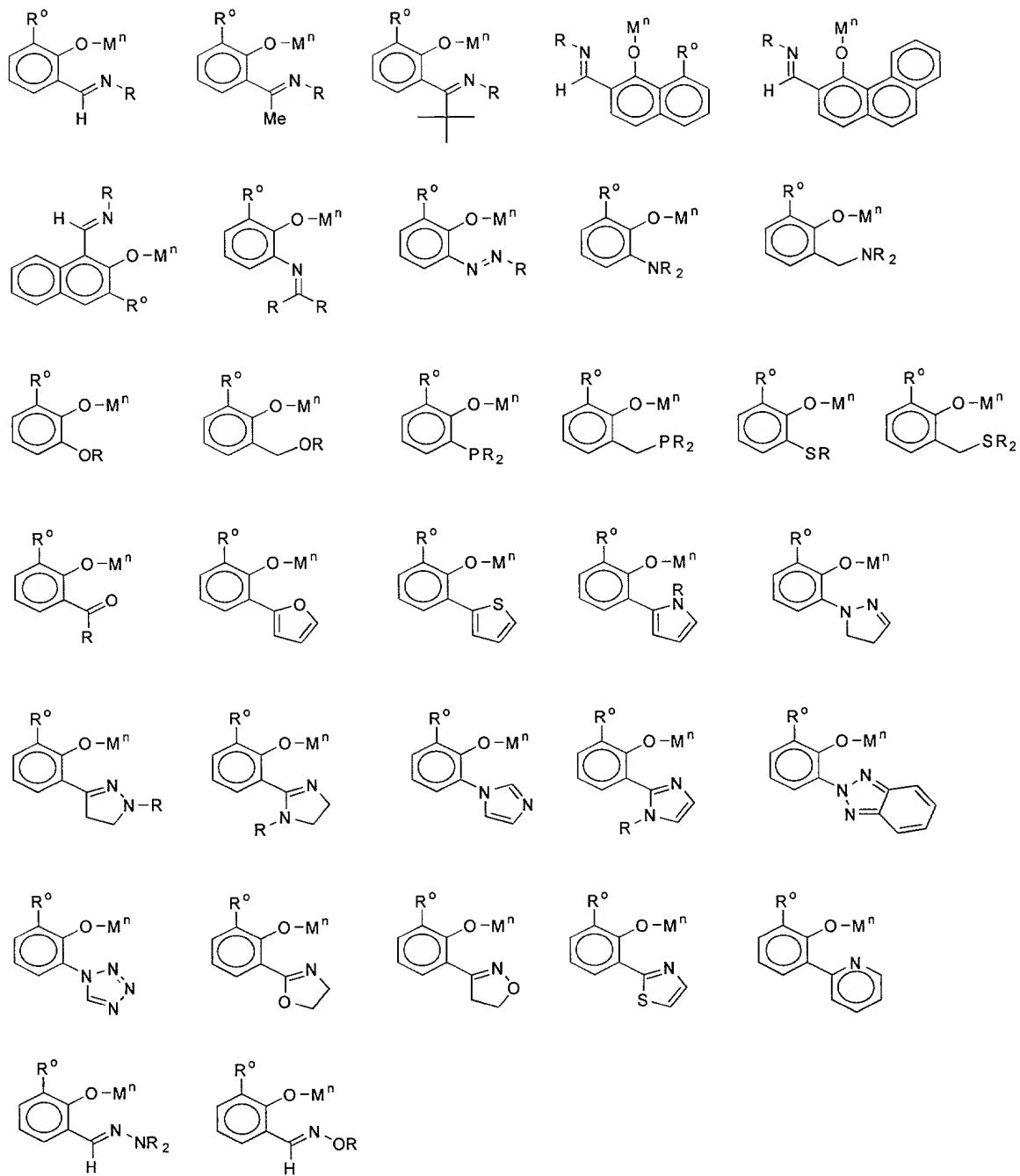
bis(*N*-*iso*-propyl-3,5-di-(1',1'-dimethylbenzyl)salicylimino)hafnium(IV) dibenzyl;
bis(*N*-*iso*-butyl-3,5-di-(1',1'-dimethylbenzyl)salicylimino)zirconium(IV) dibenzyl;
bis(*N*-*iso*-butyl-3,5-di-(1',1'-dimethylbenzyl)salicylimino)zirconium(IV) dichloride;
bis(*N*-hexyl-3,5-di-(1',1'-dimethylbenzyl)salicylimino)zirconium(IV) dibenzyl;
bis(*N*-phenyl-3,5-di-(1',1'-dimethylbenzyl)salicylimino)zirconium(IV) dibenzyl;
bis(*N*-*iso*-propyl-3,5-di-(1'-methylcyclohexyl)salicylimino)zirconium(IV) dibenzyl;
bis(*N*-benzyl-3-*t*-butylsalicylimino)zirconium(IV) dibenzyl;
bis(*N*-benzyl-3-triphenylmethylsalicylimino)zirconium(IV) dibenzyl;
bis(*N*-*iso*-propyl-3,5-di-trimethylsilylsalicylimino)zirconium(IV) dibenzyl;
bis(*N*-*iso*-propyl-3-(phenyl)salicylimino)zirconium(IV) dibenzyl;
bis(*N*-benzyl-3-(2',6'-di-*iso*-propylphenyl)salicylimino)zirconium(IV) dibenzyl;
bis(*N*-benzyl-3-(2',6'-di-phenylphenyl)salicylimino)zirconium(IV) dibenzyl;
bis(*N*-benzyl-3-*t*-butyl-5-methoxysalicylimino)zirconium(IV) dibenzyl;
bis(2-(2H-benzotriazol-2-yl)-4,6-di-*t*-amylphenoxyde)zirconium(IV) dibenzyl;
bis(2-(2H-benzotriazol-2-yl)-4,6-di-*t*-amylphenoxyde)zirconium(IV) dichloride;
bis(2-(2H-benzotriazol-2-yl)-4,6-di-*t*-amylphenoxyde)zirconium(IV)
di(bis(dimethylamide));
bis(2-(2H-benzotriazol-2-yl)-4,6-di-(1',1'-dimethylbenzyl)phenoxyde)zirconium(IV)
dibenzyl;
bis(2-(2H-benzotriazol-2-yl)-4,6-di-*t*-amylphenoxyde)titanium(IV) dibenzyl;
bis(2-(2H-benzotriazol-2-yl)-4,6-di-(1',1'-dimethylbenzyl)phenoxyde)titanium(IV)
dibenzyl;
bis(2-(2H-benzotriazol-2-yl)-4,6-di-(1',1'-dimethylbenzyl)phenoxyde)titanium(IV)
dichloride;
bis(2-(2H-benzotriazol-2-yl)-4,6-di-(1',1'-dimethylbenzyl)phenoxyde)hafnium(IV)
dibenzyl;
(*N*-phenyl-3,5-di-(1',1'-dimethylbenzyl)salicylimino)zirconium(IV) tribenzyl
(*N*-(2',6'-di-*iso*-propylphenyl)-3,5-di-(1',1'-
dimethylbenzyl)salicylimino)zirconium(IV) tribenzyl;

(*N*-(2',6'-di-*iso*-propylphenyl)-3,5-di-(1',1'-dimethylbenzyl)salicylimino)titanium(IV) tribenzyl; and (*N*-(2',6'-di-*iso*-propylphenyl)-3,5-di-(1',1'-dimethylbenzyl)salicylimino)zirconium(IV) trichloride.

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41. The catalyst system of claim 40 wherein the alumoxane is a modified alumoxane.

more transition metal compounds represented by the following formulae:



provided that M^n is bound to Q_{n-1} , and

R^5 is an aldimino, ketimino, alkoxy, α -alkoxymethyl, thioalkoxy, α -

5 thioalkoxymethyl, amino, α -aminomethyl, azo, phosphino, α -phosphinomethyl, keto

group or a pyrrole, furan, thiophene, imidazole, pyrazole, tetrazole, oxazoline, isoazole, or thiazole group,

R^o is a tertiary alkyl group or silyl group,

R is hydrogen or an alkyl, aryl, or silyl group or $-OT$ where O is oxygen and T is hydrogen or an alkyl, aryl or silyl group,

5 M is a group 4 to 10 transition metal,

n is the valence state of M, and

Q is an anionic ligand.

10 43. The catalyst system of claim 42 wherein R^o is selected from the group consisting of $-CMe_3$, $-CMe_2Et$, CEt_3 , $-CMe_2Ph$, $-CPh_3$, $-SiMe_3$, $-SiEt_3$, $-SiPh_3$, where Me is methyl, Et is ethyl and Ph is phenyl.

15 44. The catalyst system of claim 42 wherein Q is an amide, carboxylate, carbamate, thiolate, hydride or alkoxide group.

20 45. The catalyst system of claim 42 wherein Q is a halogen or an alkyl group.

46. The catalyst system of claim 42 wherein M is a group 4 metal.

25 47. The catalyst system of claim 42 wherein either the transition metal compound or the activator or both or the reaction product thereof are placed on a support.

48. The catalyst system of claim 47 wherein the support is talc; silica, magnesium chloride, alumina, silica-alumina; polyethylene, polypropylene, polystyrene; or a mixture thereof.

49. The catalyst system of claim 47 wherein prior to being combined with the transition metal compound and/or the activator the support is partially or completely dehydrated.

5 50. A catalyst system comprising one or more heteroatom substituted phenoxide group 4 to 10 transition metal or lanthanide metal compounds wherein the metal is bound to the oxygen of the phenoxide group and provided that:

- a) if more than one heteroatom substituted phenoxide is present it is not bridged to the other heteroatom substituted phenoxide,
- 10 b) if the metal is a group 4 metal, the heteroatom substituted phenoxide does not contain pyridine,
- c) if the metal is a group 4 metal then the carbon ortho to the carbon bound to the oxygen of the phenoxide may not be bound to an aldehyde or an ester, and
- 15 d) if the metal is nickel then the carbon ortho to the carbon bound to the oxygen of the phenoxide may not be bound to an imine.

51. The catalyst system of claim 50 wherein the transition metal is a group 4 metal.

20 52. The catalyst system of claim 50 wherein the transition metal is zirconium, hafnium or titanium.

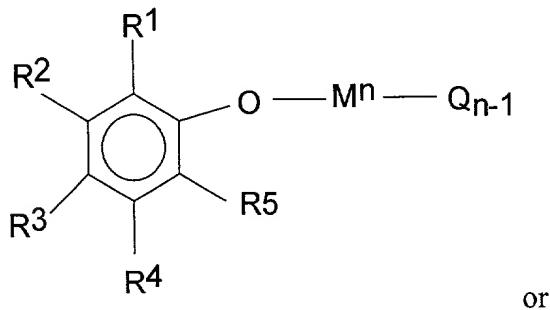
53. The catalyst system of claim 50 wherein the heteroatom substituted phenoxide transition metal compound is an iminophenoxide group 4 transition metal compound.

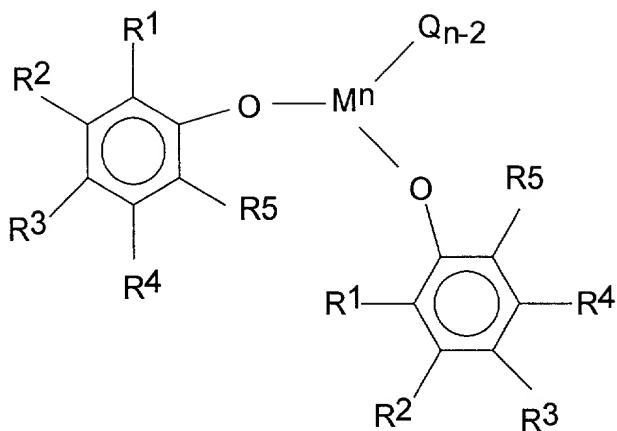
25 54. The catalyst system of claim 50 wherein the heteroatom substituted phenoxide transition metal compound is an iminophenoxydezirconium compound.

55. A process for polymerizing olefins comprising combining one or more olefins with a catalyst system comprising the reaction product of one or more activators and one or more heteroatom substituted phenoxide group 4 to 10 transition metal or lanthanide metal compounds wherein the metal is bound to the oxygen of the phenoxide group and provided that:

- a) if more than one heteroatom substituted phenoxide is present it is not bridged to the other heteroatom substituted phenoxide,
- b) if the metal is a group 4 metal, the heteroatom substituted phenoxide does not contain pyridine,
- 10 c) if the metal is a group 4 metal then the carbon ortho to the carbon bound to the oxygen of the phenoxide may not be bound to an aldehyde or an ester, and
- d) if the metal is nickel then the carbon ortho to the carbon bound to the oxygen of the phenoxide may not be bound to an imine.

15 56. The process of claim 55 wherein the heteroatom substituted phenoxide transition metal compound is represented by the of the following formulae:





wherein:

5 R^1 to R^5 may be independently hydrogen, a heteroatom containing group or a C_1 to C_{100} group provided that at least one of R^2 to R^5 is a group containing a heteroatom, any of R^1 to R^5 may or may not be bound to the metal M,

10 O is oxygen,

 M is a group 4 to 10 transition metal or a lanthanide metal,

 n is the valence state of M,

15 Q is an anionic ligand or a bond to an R group containing a heteroatom which may be any of R^1 to R^5 , and

 further provided that:

a) if M is a group 4 metal then R^5 may not be an aldehyde or an ester;

b) if M is nickel then R^5 may not be an imine;

15 c) the R^4 and R^5 groups do not form pyridine in the first formula if M is a group 4 metal; and

d) the R^4 and R^5 groups do not form pyridine in at least one ring of the second formula if M is a group 4 metal.

57. The process of claim 55 wherein the activator is an aluminum alkyl, an

20 alumoxane, a modified alumoxane, a borane, a non-coordinating anion or a mixture thereof.

58. The process of claim 56 wherein Q is a bond to any of R² to R⁵ and the R group that Q is bound to is a heteroatom containing group.

5 59. The process of claim 55 wherein the heteroatom containing group is an imime, triazole, or oxazole.

60. The process of claim 55 wherein the heteroatom in the heteroatom containing group is nitrogen and/or oxygen.

10 61. The process of claim 56 wherein the R¹ group is a C₄ to C₂₀ alkyl group.

62. The process of claim 56 wherein the R¹ group is a butyl, isobutyl, pentyl hexyl, heptyl, isohexyl, octyl, isoctyl, decyl, nonyl, or dodecyl group.

15 63. The process of claim 56 wherein two or more R groups have formed a five or six membered ring.

20 64. The process of claim 56 wherein two or more R groups have formed a multi ring system.

65. The process of claim 56 wherein M is zirconium, titanium or hafnium.

66. The process of claim 56 wherein n is 3 or 4.

25 67. The process of claim 56 wherein Q is a halogen or an alkyl group.

68. The process of claim 56 wherein Q is an amide, carboxylate, carbamate, thiolate, hydride or alkoxide group.

69. The process of claim 55 wherein the catalyst system and the olefin are reacted in the gas phase.

5 70. The process of claim 55 wherein the catalyst system and the olefin are reacted in the slurry phase.

71. The process of claim 55 wherein the catalyst system and the olefin are reacted in the slurry phase solution phase.

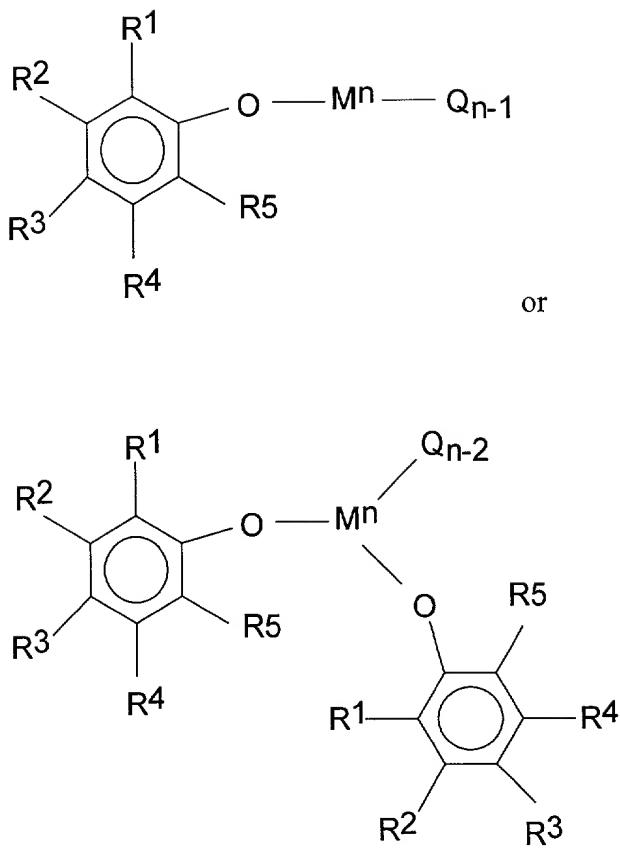
10 72. The process of claim 55 wherein the catalyst system and the olefin are reacted under high pressure.

15 73. A library of a plurality of heteroatom substituted phenoxide group 4 to 10 transition metal or lanthanide metal compounds wherein the metal is bound to the oxygen of the phenoxide group and provided that:

- a) if more than one heteroatom substituted phenoxide is present it is not bridged to the other heteroatom substituted phenoxide,
- b) if the metal is a group 4 metal, the heteroatom substituted phenoxide does not contain pyridine,
- c) if the metal is a group 4 metal then the carbon ortho to the carbon bound to the oxygen of the phenoxide may not be bound to an aldehyde or an ester, and
- d) if the metal is nickel then the carbon ortho to the carbon bound to the oxygen of the phenoxide may not be bound to an imine.

20 25 74. A process for the simultaneous parallel screening of catalysts comprising combining the library of claim 73 with one or more olefins.

75. A library of a plurality of compounds represented by the following formulae:



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wherein:

R¹ to R⁵ may be independently hydrogen, a heteroatom containing group or a C₁ to C₁₀₀ group provided that at least one of R² to R⁵ is a group containing a heteroatom,

10 R¹ to R⁵ may or may not be bound to the metal M,

O is oxygen,

M is a group 4 to 10 transition metal or lanthanide metal,

n is the valence state of M,

Q is an anionic ligand or a bond to an R group containing a heteroatom which may be any of R¹ to R⁵, and

15 further provided that:

- a) if M is a group 4 metal then R⁵ may not be an aldehyde or an ester;
- b) if M is nickel then R⁵ may not be an imine;

- c) the R^4 and R^5 groups do not form pyridine in the first formula if M is a group 4 metal; and
- d) the R^4 and R^5 groups do not form pyridine in at least one ring of the second formula if M is a group 4 metal.

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76. A process for the simultaneous parallel screening of catalysts comprising combining the library of claim 75 with one or more olefins.

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Abstract

This invention relates to a catalyst system comprising an activator and one or more heteroatom substituted phenoxide group 4 to 10 transition metal or lanthanide metal compounds wherein the metal is bound to the oxygen of the phenoxide group and provided that:

- a) if more than one heteroatom substituted phenoxide is present it is not bridged to the other heteroatom substituted phenoxide,
- b) if the metal is a group 4 metal, the heteroatom substituted phenoxide does not contain pyridine,
- c) if the metal is a group 4 metal then the carbon adjacent to the carbon bound to the oxygen of the phenoxide may not be bound to an aldehyde or an ester, and
- d) if the metal is nickel then the carbon adjacent to the carbon bound to the oxygen of the phenoxide may not be bound to an imine.

The activator may be an aluminum alkyl, an alumoxane, a modified alumoxane, a non-coordinating anion, a borane or a mixture thereof.

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